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**Original Paper** 

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# Identification and geochemical significance of unusual C<sub>24</sub> tetracyclic terpanes in Shahejie Formation source rocks in the Bozhong subbasin, Bohai Bay Basin



## Ning Wang <sup>a, b, \*</sup>, Yao-Hui Xu <sup>a, b</sup>, Fei-Long Wang <sup>c</sup>, Yan Liu <sup>a</sup>, Qian Huang <sup>a</sup>, Xing Huang <sup>a</sup>

<sup>a</sup> Key Laboratory of Exploration Technologies for Oil and Gas Resources, Ministry of Education, Yangtze University, Wuhan, 430100, China <sup>b</sup> Hubei Key Laboratory of Petroleum Geochemistry and Environment (Yangtze University), Wuhan, 430100, China

<sup>c</sup> Bohai Petroleum Research Institute, Tianjin Branch, CNOOC (China) Ltd., Tianjin, 300452, China

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#### ABSTRACT

 $C_{24}$  tetracyclic terpanes are common compounds in source rocks and crude oils, and  $C_{24}$  17,21secohopane is the most common and widely used source-related indicator. In this study, three unusual  $C_{24}$  tetracyclic terpanes were detected on the m/z 191 chromatogram of saturated hydrocarbons in the Shahejie Formation source rocks in the Bozhong subbasin. Based on the mass spectra characteristics, diagnostic ion fragments, retention time and comparisons with published literature, three unusual  $C_{24}$ tetracyclic terpanes were identified as  $10\beta(H)$ -des-A-oleanane,  $10\beta(H)$ -des-A-lupane and  $C_{24}$  des-Ahopane. To the best of our knowledge, this is the first study to detect and publicly report these three compounds in source rock samples from the Shahejie Formation of the Bozhong subbasin, Bohai Bay Basin. The results indicated that  $10\beta(H)$ -des-A-oleanane and  $10\beta(H)$ -des-A-lupane likely originated from terrestrial angiosperms, while  $C_{24}$  des-A-hopane likely originated from prokaryotic organisms. Terrestrial angiosperms provide the material basis for the generation of compounds A and B, and the distribution and concentration of these two compounds are affected by thermal maturity. In the low maturity stage ( $0.5\% < R_0 < 0.7\%$ ), compounds A and B are relatively enriched in the source rocks.

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#### 1. Introduction

 $C_{24}$  tetracyclic terpanes are widely found in sediments and crude oils (Peters et al., 2005; Bao et al., 2018). The formation of  $C_{24}$ tetracyclic terpanes in sediments occurs mainly in two ways: one is related to the cleavage of the C-17 and C-21 bonds on the E-rings of hopanes or hopanoids caused by microbial activity or thermal effects (Ekweozor et al., 1981; Connan et al., 1986), and the other is related to the decyclization of the E-rings or A-rings of terrestrial triterpenes mediated by microorganisms (Woolhouse et al., 1992; Samuel et al., 2010; Bao et al., 2018). The tetracyclic terpane formed by the former mechanism is also known as 17,21-secohopane (Trendel et al., 1982; Lu et al., 2009). Most of these compounds occur as single  $C_{24}$  17,21-secohopane compounds (Hu, 1991; Zhang

\* Corresponding author. Key Laboratory of Exploration Technologies for Oil and Gas Resources, Ministry of Education, Yangtze University, Wuhan, 430100, China. *E-mail address:* cupwangning@outlook.com (N. Wang). et al., 2003) or as a C24-C27 17,21-secohopane homolog series in sediments and crude oils (Aquino et al., 1983; Woolhouse et al., 1992). In addition, the C<sub>28</sub>-C<sub>30</sub> 17,21-secohopane homolog series was also detected in evaporitic sediments in the Bohai Bay Basin (Lu et al., 2009). In general, C24 17,21-secohopane is the most typical compound in these series and is especially enriched in carbonate and evaporative sedimentary environments (Palacas et al., 1984; Lu et al., 2009; Chattopadhyay and Dutta, 2014). However, C24 17,21secohopane is also widely found in source rocks deposited in various sedimentary environments, such as marine/lacustrine shale/mudstone and terrestrial coal seam environments (Ekweozor et al., 1981; Woolhouse et al., 1992; Kruge et al., 1996; Bao et al., 1999; Samuel et al., 2010). This kind of C<sub>24</sub> tetracyclic terpanes (C24 17,21-secohopane) has been widely used in oil-source correlations (Woolhouse et al., 1992; Peters et al., 2005; Samuel et al., 2010), and C<sub>24</sub> 17,21-secohopane (C<sub>24</sub>TeT)/hopane, C<sub>24</sub>TeT/C<sub>23</sub> tricyclic terpane (TT) and C24TeT/C26TT ratios are common sourcerelated indicators (Peters et al., 2005).

With the in-depth study of tetracyclic terpanes, some novel C<sub>24</sub>

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tetracyclic terpanes have been detected and identified in crude oils in addition to C<sub>24</sub> 17,21-secohopane. These C<sub>24</sub> tetracyclic terpanes are believed to be formed by the decyclization of the E-ring or Aring of terrestrial triterpenes mediated by microorganisms (Woolhouse et al., 1992; Samuel et al., 2010; Bao et al., 2018). These A/E-ring degraded tetracyclic terpanes usually contain des-Aoleanane. des-A-lupane, des-A-ursane, and des-E-lupane (Woolhouse et al., 1992; Chattopadhyay and Dutta, 2014). Woolhouse et al. (1992) first detected a series of ring-degraded  $C_{24}$ tetracyclic terpanes in crude oils and tentatively identified them as  $10\beta(H)$ -des-A-oleanane,  $10\beta(H)$ -des-A-lupane, and  $10\beta(H)$ -des-Aursane. Subsequently,  $10\beta(H)$ -des-A-oleanane was also detected in organic-rich Holocene sediment extracts (Bechtel et al., 2007), Tertiary crude oils from the Niger Delta (Samuel et al., 2010) and Eocene coal-bearing sediments from India (Chattopadhyay and Dutta, 2014). These novel C<sub>24</sub> tetracyclic terpanes have also been detected in some domestic crude oil and source rock extracts since 2014, although all of these sources were from the Pearl River Mouth Basin (Ma et al., 2014; Zhu et al., 2015; Bao et al., 2018). To date, few reports have focused on these novel tetracyclic terpane compounds in the source rocks of other sedimentary basins in China.

In this study, four  $C_{24}$  tetracyclic terpanes, including three unusual  $C_{24}$  tetracyclic terpanes, were detected in the Shahejie Formation source rocks in the Bozhong subbasin, Bohai Bay Basin. These unusual  $C_{24}$  tetracyclic terpanes have not been previously reported in the study area. In this study, we systematically analyzed the distribution patterns of these three unusual  $C_{24}$  tetracyclic terpanes in source rocks to investigate possible factors affecting the unusual  $C_{24}$  tetracyclic terpanes generated and concentrated in source rocks and their potential geochemical significance. The results are expected to help geochemists better understand the geochemical significance of these unusual  $C_{24}$  tetracyclic terpanes.

#### 2. Geological setting

The Bohai Bay Basin is a Meso-Cenozoic extensional rift basin located in northeastern China (Yang and Xu, 2004; Xu et al., 2019) that experienced two major tectonic movements during the Cenozoic (Gong et al., 2007) and formed seven subbasins: Jiyang, Linqing, Jizhong, Huanghua, Liaohe, Liaodong Bay and Bozhong (Fig. 1a). The Bozhong subbasin is one of the important petroleumproducing areas and favorable exploration areas in the Bohai Bay Basin (Yin et al., 2020). In recent years, an increasing number of oil and gas resources have been discovered in the Bozhong subbasin. A 100 billion cubic meter gas field (BZ19–6) was successfully discovered in the Bozhong subbasin in 2019 (Xu et al., 2019). More recently, another large-scale oil and gas (BZ13–2) field was discovered in the Bozhong subbasin, which has proven geological reserves of 100 million tons of oil and gas equivalent (Shi et al., 2021).

The Bozhong subbasin contains two main sags and several uplifts in the southwest, namely, the Bozhong sag, Huanghekou sag, Shaleitian uplift, Shijiutuo uplift, Bonan lower uplift and Bodong lower uplift (Fig. 1b). A succession of fluvial-lacustrine sediments, including the Kongdian, Shahejie and Dongying formations, was deposited during the Paleogene in the Bozhong and Huanghekou sags (Fig. 1c). Among them, the Shahejie Formation can be divided into four members from top to bottom: E<sub>2</sub>s<sub>1</sub>, E<sub>2</sub>s<sub>2</sub>, E<sub>2</sub>s<sub>3</sub> and E<sub>2</sub>s<sub>4</sub>, E<sub>2</sub>s<sub>1</sub> and E<sub>2</sub>s<sub>3</sub> are mainly composed of mudstones and silty mudstones, which are important source rocks in the Bozhong subbasin (Hao et al., 2011). Previous studies indicated that terrestrial higher plants, phytoplankton and algae were the main biological sources of E<sub>2</sub>s<sub>1</sub> and E<sub>2</sub>s<sub>3</sub> (Hao et al., 2011; Yin et al., 2020), and E<sub>2</sub>s<sub>1</sub> and E<sub>2</sub>s<sub>3</sub> were deposited in highly saline and fresh lacustrine environments, respectively (Hao et al., 2011; Jiang et al., 2019).

#### 3. Samples and methods

A total of 25 mudstone/shale core samples were collected from the Bozhong and Huanghekou sags, with nineteen samples taken from  $E_2s_3$  at depths from ~3356 m to ~3684 m and the remaining six samples taken from  $E_2s_1$  at depths from 3034 m to 3297.5 m (Table 1).

All samples were washed using deionized water and then crushed to < 100 mesh after drying. The powder samples were extracted using Soxhlet extraction for 48 h with dichloromethane. Isolated extracts were fractionated by column chromatography after the asphaltene was precipitated by adding excessive petroleum ether. The remaining saturated, aromatic and resin fractions were separated by a column filled with active silica gel and alumina with appropriate elution solvents. The elution solvents were *n*-hexane, dichloromethane:petroleum ether (2:1) and dichloromethane:methyl alcohol (93:7).

Gas chromatography-mass spectrometry (GC-MS) analyses of the saturated hydrocarbons were conducted on an Agilent-6890N gas chromatograph coupled with an Agilent-5975 mass spectrometer. The Agilent-6890N gas chromatograph was equipped with an HP-5MS fused silica capillary column (30 m  $\times$  0.25 mm  $\times$  0.25 µm), and the carrier gas was helium. The initial temperature of the GC oven for saturated analysis was 50 °C (held for 1 min); then, the temperature was increased to 120 °C at a rate of 20 °C per minute and finally increased to 310 °C at a rate of 20 °C per minute and held for 25 min.

#### 4. Results and discussion

#### 4.1. Basic geological information of the studied samples

Before discussing the main factors controlling the abundance of these three unusual compounds in geological samples, it is necessary to investigate the basic geological information of the studied  $E_2s_1$  and  $E_2s_3$  source rock samples, such as the thermal maturity, sedimentary environment and organic matter sources.

The maximum pyrolysis temperatures ( $T_{max}$ ) of the  $E_2s_1$  and  $E_2s_3$  samples vary from 436 °C to 447 °C and 429 °C to 447 °C, respectively (Table 1), suggesting a low maturity to maturity stage. The sterane isomer ratios  $C_{29}\alpha\alpha\alpha$ -20S/(20S+20R) and  $C_{29}\alpha\beta\beta/(\alpha\beta\beta+\alpha\alpha\alpha)$ , which represent the degree of biological configuration transformation to geological configuration, are widely used to evaluate the thermal maturity of source rocks (Seifert and Moldowan, 1986; Peters et al., 2005). The  $C_{29}\alpha\alpha\alpha$ -20S/(20S+20R) and  $C_{29}\alpha\beta\beta/(\alpha\beta\beta+\alpha\alpha\alpha)$  ratios of  $E_2s_1$  are in the range of 0.29–0.45 (mean of 0.41) and 0.30–0.35 (mean of 0.32) (Table 1), respectively, suggesting a low maturity stage. The  $C_{29}\alpha\alpha\alpha$ -20S/(20S+20R) and  $C_{29}\alpha\beta\beta/(\alpha\beta\beta+\alpha\alpha\alpha)$  values of  $E_2s_3$  are higher than those of the  $E_2s_1$  samples, indicating a low maturity-maturity stage (Seifert and Moldowan, 1986; Peters et al., 2005).

The ratio of pristane to phytane (Pr/Ph) is often used to evaluate redox conditions in bottom water (Didyk et al., 1978). Generally, anoxic conditions are conducive to the formation of Ph, while oxic conditions promote the formation of Pr, resulting in a lower Pr/Ph ratio under anoxic conditions (Didyk et al., 1978; Shanmugam, 1985). The  $E_{2S_1}$  source rock samples exhibit relatively lower Pr/Ph values (0.84–1.4) than those of the  $E_{2S_3}$  source rock samples (1.13–2.12), suggesting that  $E_{2S_1}$  was deposited under more anoxic conditions than  $E_{2S_3}$  (Table 1). The high abundance of gammacerane in source rock samples is usually associated with high salinity water conditions (Moldowan et al., 1985; Summons et al., 2008). Except for the two samples, most of the gammacerane indices (gammacerane/C<sub>30</sub> hopane) of the  $E_{2S_3}$  source rock samples are in the range of 0.03–0.07 (Table 1), indicating freshwater conditions.



Fig. 1. (a) The location of the Bohai Bay Basin and its sub-basin classification (according to Liu et al., 2016); (b) Simplified structural map showing the tectonic units and sampling well in the Bozhong sub-basin (modified after Yin et al., 2020); (c) Generalized stratigraphic column of the Paleogene-Quaternary in the Bohai Bay Basin (modified after Liu et al., 2016).

The  $E_{2s_1}$  source rock samples have wide ranges of gammacerane index values, suggesting a great variation in water salinity.

The relative abundances of  $C_{27}-C_{29}$  regular steranes are effective indices to reflect the origin of organic matter, especially the contribution of eukaryotes (Gaskell and Eglinton, 1973; Peters, 2005). A previous study proposed that  $C_{27}$  regular steranes in sediments are related to the contribution of phytoplankton or algae while  $C_{29}$  regular steranes mainly originate from terrestrial higher plants (Volkman et al., 2005). The average proportion of  $C_{27}$  regular

steranes among the C<sub>27</sub>–C<sub>29</sub> regular steranes is ~38% in the E<sub>2</sub>s<sub>3</sub> source rock samples, which is slightly lower than that of the C<sub>29</sub> regular steranes (Table 1), suggesting the mixed input of phytoplankton, algae and terrestrial higher plants to E<sub>2</sub>s<sub>3</sub> organic matter. In addition, 4-methylsteranes are also detected in the E<sub>2</sub>s<sub>3</sub> source rock samples, which are believed to have originated from dinoflagellates in the Bohai Bay Basin (Zhang et al., 2005; Hao et al., 2011). However, the E<sub>2</sub>s<sub>1</sub> samples contain a high abundance of C<sub>29</sub> regular steranes compared to C<sub>27</sub> and C<sub>28</sub> regular steranes

The relevant biomarker parameters of the studied  $E_2s_1$  and  $E_2s_3$  samples in the study area.

No.	Well	Depth, m	Stratum	$T_{max}$ , °C	$Ga/C_{30}H$	C <sub>27</sub> %	C <sub>28</sub> %	C <sub>29</sub> %	Ι	II	III	IV	C <sub>30</sub> H/C <sub>23</sub> TT	A/C <sub>23</sub> TT	B/C <sub>23</sub> TT	C/C <sub>23</sub> TT	A%	B%	C%
1	BZ-a1	3356	E <sub>2</sub> s <sub>3</sub>	439	0.43	44.8	18.4	36.8	1.37	0.23	0.31	0.41	17.5	0.12	0.00	0.06	65.4	0.0	34.6
2	BZ-a1	3357	$E_2s_3$	434	0.42	45.6	16.7	37.8	1.41	0.27	0.32	0.42	16.4	0.08	0.00	0.04	63.4	3.8	32.8
3	BZ-a1	3359.2	$E_2s_3$	437	0.05	38.6	20.2	41.2	1.24	0.12	0.51	0.42	67.2	0.75	0.05	0.50	57.9	3.8	38.4
4	BZ-a1	3412.6	$E_2s_3$	434	0.05	37.5	17.0	45.4	1.42	0.14	0.37	0.48	58.3	0.38	0.03	0.07	78.7	6.2	15.1
5	BZ-a1	3414.2	$E_2s_3$	435	0.04	36.0	19.7	44.2	1.61	0.18	0.42	0.47	61.6	0.50	0.02	0.12	77.7	3.6	18.7
6	BZ-a1	3416.4	$E_2s_3$	431	0.05	36.6	21.3	42.2	1.63	0.24	0.42	0.50	78.7	0.98	0.04	0.13	85.0	3.7	11.3
7	BZ-a1	3419.9	$E_2s_3$	436	0.05	36.0	18.8	45.3	1.55	0.24	0.43	0.47	63.3	0.83	0.03	0.07	88.7	3.6	7.6
8	BZ-a1	3422.7	$E_2s_3$	1	0.05	35.9	15.5	48.5	1.59	0.21	0.44	0.45	59.1	0.72	0.03	0.10	84.6	3.6	11.8
9	BZ-a1	3423.9	$E_2s_3$	436	0.05	36.7	18.1	45.2	1.66	0.24	0.44	0.47	44.6	0.87	0.03	0.09	88.3	2.9	8.8
10	BZ-a1	3426.4	$E_2s_3$	436	0.05	34.4	22.6	43.0	1.13	0.30	0.45	0.52	52.1	1.36	0.04	0.05	93.9	2.5	3.6
11	BZ-a1	3665.2	$E_2s_3$	447	0.05	45.3	22.0	32.7	1.79	0.63	0.53	0.54	14.3	0.84	0.00	0.04	95.2	0.0	4.8
12	BZ-b1	3543.8	$E_2s_3$	435	0.06	37.1	21.3	41.6	1.14	0.09	0.41	0.43	29.4	0.42	0.00	0.34	55.7	0.0	44.3
13	BZ-b1	3545.7	$E_2s_3$	439	0.07	38.8	20.9	40.4	1.17	0.10	0.45	0.41	49.5	0.44	0.05	0.29	56.8	6.6	36.6
14	BZ-b1	3657.4	$E_2s_3$	429	0.07	39.6	22.4	38.0	1.28	0.17	0.42	0.49	16.7	0.23	0.00	0.07	75.6	0.0	24.4
15	BZ-b1	3658.8	$E_2s_3$	440	0.07	37.5	25.8	36.8	1.66	0.17	0.38	0.50	25.0	0.24	0.00	0.09	73.4	0.0	26.6
16	BZ-b1	3660.6	$E_2s_3$	441	0.05	38.0	24.0	38.0	1.44	0.17	0.43	0.48	23.2	0.27	0.00	0.09	75.3	0.0	24.7
17	BZ-b1	3677.5	$E_2s_3$	437	0.03	33.5	27.3	39.1	1.69	0.31	0.40	0.53	65.0	0.27	0.00	0.09	75.7	0.0	24.3
18	BZ-b1	3683.5	$E_2s_3$	443	0.05	35.6	25.0	39.5	2.12	0.19	0.42	0.49	27.9	0.90	0.00	0.11	88.8	0.0	11.2
19	BZ-b1	3684.5	$E_2s_3$	442	0.03	38.1	24.0	38.0	1.66	0.18	0.43	0.52	26.2	0.37	0.00	0.09	80.5	0.0	19.5
20	BZ-c1	3276.65	$E_2s_1$	441	0.42	26.4	19.3	54.3	1.70	0.80	0.47	0.33	22.1	6.20	0.70	0.22	87.1	9.8	3.1
21	BZ-c2	3276	$E_2s_1$	447	0.75	37.6	22.0	40.5	0.85	0.54	0.41	0.35	28.3	2.95	0.45	0.05	85.7	12.9	1.4
22	BZ-c2	3281.4	$E_2s_1$	441	0.10	28.4	17.5	54.1	0.84	0.71	0.45	0.31	9.6	0.29	0.03	0.10	70.1	6.6	23.3
23	BZ-c2	3287	$E_2s_1$	445	0.06	36.1	19.7	44.2	1.40	0.18	0.38	0.33	90.4	2.48	0.18	0.82	71.3	5.1	23.6
24	BZ-c2	3297.5	$E_2s_1$	444	0.06	19.8	16.3	63.9	1.33	0.83	0.46	0.30	23.8	5.12	0.33	0.24	90.1	5.8	4.1
25	BZ-d1	3035	$E_2s_1$	436	0.36	54.3	18.6	27.1	1.30	0.33	0.29	0.32	6.3	0.21	0.03	0.04	76.0	10.8	13.2

Note: Ga = gammacerane;  $C_{30}H = C_{30}$ -hopane;  $C_{27}\% = C_{27}$  regular steranes/ $(C_{27}+C_{28}+C_{29})$  regular steranes; I = pristane/phytane; II = oleanane/(oleanane+C\_{29} hopane); III =  $C_{29}\alpha\alpha\alpha$ -20S/(20S+20R); IV =  $C_{29}\alpha\beta\beta/(\alpha\beta\beta+\alpha\alpha\alpha)$ .

(Table 1), indicating a significant input of terrestrial higher plants. The dominant contribution of higher plants to  $E_2s_1$  samples is also confirmed by the relatively high content of oleanane (Table 1). The relatively high abundance of hopane series biomarkers in both the  $E_2s_1$  and  $E_2s_3$  samples suggests a large amount of prokaryotic organism input. In summary, the organic matter in the  $E_2s_1$  and  $E_2s_3$  samples was derived from a variety of organisms, including primary producers, such as algae/phytoplankton, terrestrial higher plants and prokaryotic organisms, e.g., bacteria. However, the contribution ratios of different kinds of organisms to  $E_2s_1$  and  $E_2s_3$  organic matter are different. The  $E_2s_1$  source rocks are dominated by terrigenous higher plant inputs, while the  $E_2s_3$  source rocks are characterized by typical mixed organic matter sources.

## 4.2. Occurrence and identification of unusual $C_{24}$ tetracyclic terpanes

Three uncommon compounds in geological samples were detected on the m/z 191 chromatogram of saturated hydrocarbons from the studied samples, which are termed A, B and C (Fig. 2). In addition, C<sub>24</sub> 17,21-secohopane (C<sub>24</sub>TeT), which is widely present in sedimentary organic matter, can also be clearly identified in all E<sub>2</sub>s<sub>3</sub> and E<sub>2</sub>s<sub>1</sub> samples. Compounds A, B and C elute midway between tricyclic terpanes C<sub>24</sub> and C<sub>25</sub> in the order of increasing elution time (Fig. 2). The relative contents of compound A, which ranges from ~57% to ~95%, is dominant among the three C<sub>24</sub> tetracyclic terpanes (Table 1). In some  $E_2s_1$  source rock samples, the relative abundance of compound A is even higher than that of  $C_{19}-C_{29}TT$  (Fig. 2). Compound B is present in relatively lower abundance in the m/z 191 chromatogram, and its concentration is even below the detection level in some E<sub>2</sub>s<sub>3</sub> source rock samples. However, the relative abundance of compound C varies greatly in different samples. In some studied samples, its relative content is as high as 44% among the three compounds, while in other source rock samples, its content is lower than 2% (Table 1).

The molecular structures of compounds A, B and C were further determined by analysis of ion fragments and retention time and comparison with published literature, which is a common method of determining the molecular structure of unknown compounds (Li et al., 2015). The representative mass spectra of the three  $C_{24}$  tetracyclic terpanes after subtracting the background mass spectra are shown in Fig. 3.

For compound A, the base peak and molecular ion peak occur at m/z 191 and 330, respectively, and relatively high contents of ion fragments occur at m/z 109, 177 and 206 (Fig. 3a), indicating that this is a  $C_{24}$  tetracyclic terpane with a molecular formula of  $C_{24}H_{42}$ . Furthermore, the mass spectra of compound A in this study are almost identical to the published mass spectra of  $10\beta(H)$ -des-Aoleanane (Woolhouse et al., 1992; Samuel et al., 2010; Bao et al., 2018). The retention time of compound A is also identical to that of 10β(H)-des-A-oleanane reported in previous studies (Woolhouse et al., 1992; Samuel et al., 2010; Bao et al., 2018). Thus, compound A is classified as  $10\beta(H)$ -des-A-oleanane based on the analysis of the retention time and mass spectra. The mass spectra of compound B present a relatively high content of the characteristic ion fragment at m/z 287, which is obviously different from that of compound A (Fig. 3b). The presence of this fragment is usually associated with the cleavage of an isopropyl from the terminal five-membered ring (Woolhouse et al., 1992). In addition, the base peak and molecular ion peak are at m/z 123 and 330, respectively. All of these mass spectral characteristics of compound B are similar to those of  $10\beta(H)$ -des-A-lupane reported in previous literature (Trendel et al., 1989; Woolhouse et al., 1992; Bao et al., 2018), indicating that compound B can be identified as  $10\beta(H)$ -des-A-lupane. This finding is also supported by the identical elution order and relative retention time of compound B and  $10\beta(H)$ -des-A-lupane (Bao et al., 2018). Compound C also has an ion fragment at m/z 287 and a molecular ion at m/z 330, indicating that this is a C<sub>24</sub> tetracyclic terpane with an isopropyl substituent group in its molecular structure. However, the different base peak ion fragments and retention times indicate that the molecular structure of compound C is different from that of compound B. The identical mass spectrum to compound C has been reported by Zhu et al. (2015), although the molecular structure is still unknown. Subsequently, Bao et al. (2018)



**Fig. 2.** Representative m/z 191 chromatograms of saturated hydrocarbons showing the distribution of novel tetracyclic terpanes from the  $E_2s_1$  and  $E_2s_3$  in the Bozhong sub-basin. TT = tricyalic terpane; TeT = tetracyclic terpane.

detected tetracyclic terpane  $C_{24}$  in crude oils from the Pearl River Mouth Basin, and its mass spectrum was identical to that of compound C in this study and the unknown structural compound discovered by Zhu et al. (2015). Based on the diagnostic ion fragment at m/z 287 and 191, Bao et al. (2018) identified this compound as  $C_{24}$  des-A-hopane. In this study, the research results of Bao et al. (2018) are accepted; thus, compound C is tentatively identified as  $C_{24}$  des-A-hopane. In summary, compounds A, B and C are classified as  $10\beta$ (H)-des-A-oleanane,  $10\beta$ (H)-des-A-lupane and  $C_{24}$  des-Ahopane based on diagnostic ion fragments and the retention time.

#### 4.3. Origin of the unusual C<sub>24</sub> tetracyclic terpanes

The above analyses show that compounds A and B are  $10\beta$ (H)des-A-oleanane and  $10\beta$ (H)-des-A-lupane, respectively, indicating that they are degraded products of oleanane and lupane. In addition, higher A/C<sub>23</sub>TT and B/C<sub>23</sub>TT ratios are generally observed in the E<sub>2</sub>s<sub>1</sub> source rocks, where terrigenous higher plants are important biological sources, while relatively lower A/C<sub>23</sub>TT and B/C<sub>23</sub>TT ratios are observed in the E<sub>2</sub>s<sub>3</sub> source rocks, where mixed organic matter is an important source (Table 1). Based on the above analyses, we propose that the precursors of compounds A and B may be



Fig. 3. Representative mass spectra of compounds A, B and C, and molecular structures of compounds A and B.

the same as those of oleanane and lupane. Previous studies have suggested that oleanane and lupane originate from angiosperms; thus, they are widely used as source indicators of terrigenous angiosperms (Ekweozor and Telnae, 1990; Peters et al., 2005). The possibility that compounds A and B may originate from terrestrial angiosperms is therefore proposed in this study.

A biosynthetic pathway for the formation of compounds A and B further supports that these two unusual tetracyclic terpanes originate from terrestrial angiosperms (Fig. 4).  $\beta$ -amyrin and betulin are two natural constituents of terrigenous angiosperms that can be oxidized to corresponding ketones ( $\beta$ -amrone and lupanone) under certain conditions (reaction processes of 1, 5, and 6 in Fig. 4; Corbet et al., 1980; Samuel et al., 2010). Under anoxic conditions with the participation of microorganisms, the A-ring of these ketones are

degraded and form corresponding 3,4-seco-triterpenoid acids (reaction processes of 2 and 7 in Fig. 4; Trendel et al., 1989; Huang et al., 2008). This reaction process can also occur through photochemical alteration reactions under nonreducing conditions (Simoneit et al., 2009). The reaction of A ring degradation and 3,4seco-triterpenoid acid formation may have occurred before deposition in the precursor plant material or during the deposition stage because 3,4-seco-triterpenoid acids have been detected in immature sediments and some higher plants (Devon and Scott, 1975; Schmitter et al., 1981; Huang et al., 1996). With increasing burial depth, 3,4-seco-olean-12-en-3-oic acid and 3,4-seco-lup-3-oic acid can further deacidify and finally form 10 $\beta$ (H)-des-A-oleanane and 10 $\beta$ (H)-des-A-lupane (reaction processes of 3, 4, and 8 in Fig. 4; Samuel et al., 2010). The intermediate products of this process have



Fig. 4. Proposed formation pathway of compounds A and B (modified after Corbet et al., 1980; Trendel et al., 1989; Huang et al., 2008; Samuel et al., 2010).

been detected in sediments and crude oils, which further suggests that this biosynthetic pathway may occur under geological conditions (Schmitter et al., 1981; Huang et al., 1996). Based on the above analysis, compounds A and B can be generated by the transformation of related ketones and alcohols in higher angiosperms. The proposed formation pathways of compounds A and B are shown in Fig. 4.

Compound C ( $C_{24}$  des-A-hopane) has no obvious correlation with compounds A and B, indicating that they may have different biological sources. However, the relative content of compound C (calculated as C/C<sub>23</sub>TT) in the studied samples exhibited a positive relationship with the relative abundance of hopane (calculated as C<sub>30</sub> hopane/C<sub>23</sub>TT) (Fig. 5d). In addition, Bao et al. (2018) proposed that compound C formed under certain sedimentary and diagenetic conditions by the degradation of the A-ring of hopane or its precursors, such as diploptene and diplopterol. Considering that hopane series compounds originate from prokaryotic organisms (Ourisson et al., 1979; Peters et al., 2005; Zhao et al., 2021), a similar origin of compound C is inferred in this study.

## 4.4. Possible factors affecting the unusual $C_{24}$ tetracyclic terpanes generated and concentrated in source rocks

The relative contents of these three tetracyclic terpanes vary greatly in different  $E_2s_1$  and  $E_2s_3$  source rock samples. Generally, the concentration of a certain compound in sediments is related to three factors: biological origin, sedimentary environment and thermal maturity. A positive relationship between the relative content of compound A and oleanane has been observed in previous crude oil studies (Samuel et al., 2010; Zhu et al., 2015; Bao et al.,

contents of the three compounds and the Pr/Ph ratios of the  $E_{2S_1}$  and  $E_{2S_3}$  source rock samples is shown graphically in Fig. 6a–c. Overall, the relative contents of the three compounds in source rock

ratios, indicating that the redox condition has a limited effect on the concentration of these three compounds. In addition, the relative contents of these three compounds are also independent of the variation in the gammacerane index ( $Ga/C_{30}$  hopane) (Fig. 6d–f). The formation pathways of compounds A and B indicate that

samples do not show systematic variations with changes in Pr/Ph

The formation pathways of compounds A and B, indicate that the degree of thermal maturity plays an important role in



nisms ronment may be more conducive to the formation and preservation imilar of compounds A, B and C (Samuel et al., 2010; Zhu et al., 2015; Bao et al., 2018). However, the results of this study do not seem to support this conclusion. The relationship between the relative

A previous study suggested that an oxidized depositional envi-

2018). Based on this, researchers propose that the input of higher

plants is significantly correlated with the content of compound A in

samples (Samuel et al., 2010; Zhu et al., 2015; Bao et al., 2018). In

this study, the values of  $A/C_{23}TT$  in the  $E_2s_1$  and  $E_2s_3$  source rocks

range from 0.08 to 1.36 and 0.29 to 6.2, respectively, indicating that

the relative content of compound A in the  $E_{2S_1}$  samples is higher than that in the  $E_{2S_3}$  source rocks. This is consistent with the higher

input of terrigenous higher plants in the E<sub>2</sub>s<sub>1</sub> source rocks. How-

ever, the A/C<sub>23</sub> TT, B/C<sub>23</sub> TT and C/C<sub>23</sub> TT ratios have no significant

correlation with the oleanane contents in most of the E<sub>2</sub>s<sub>1</sub> and E<sub>2</sub>s<sub>3</sub>

samples (Fig. 5a-c), suggesting that there are other factors

affecting the contents of compounds A and B in addition to

terrigenous higher plant input. However, the positive correlation between  $C/C_{23}$  TT and  $C_{30}H/C_{23}$ TT ratios indicates that the con-

centration of compound C may be related to  $C_{30}H$ .

**Fig. 5.** Crossplot of A/C<sub>23</sub>TT versus OL/(OL+C<sub>29</sub>H) (a), B/C<sub>23</sub>TT versus OL/(OL+C<sub>29</sub>H) (b), C/C<sub>23</sub>TT versus OL/(OL+C<sub>29</sub>H) (c) and C/C<sub>23</sub>TT versus C<sub>30</sub>H/C<sub>23</sub>TT (d) of E<sub>2</sub>s<sub>1</sub> and E<sub>2</sub>s<sub>3</sub> source rock samples in study area. OL = oleanane; C<sub>29</sub>H = C<sub>29</sub> hopane.



**Fig. 6.** Crossplot of A/C<sub>23</sub>TT versus Pr/Ph (a), B/C<sub>23</sub>TT versus Pr/Ph (b), C/C<sub>23</sub>TT versus Pr/Ph (c), A/C<sub>23</sub>TT versus Ga/C<sub>30</sub>H (d), B/C<sub>23</sub>TT versus Ga/C<sub>30</sub>H (e) and C/C<sub>23</sub>TT versus Ga/C<sub>30</sub>H (f) of E<sub>2</sub>s<sub>3</sub> and E<sub>2</sub>s<sub>1</sub> source rock samples in study area.

controlling the formation of the two compounds. Two important pathways in the formation of compounds A and B, the degradation of the A ring of  $\beta$ -amrone and lupanone and the deacidification of 3,4-seco-olean-12-en-3-oic acid and 3,4-seco-lup-3-oic acid, are all controlled by the degree of thermal maturity. Thermal maturity is an important factor controlling the concentration of compounds A and B, which is also supported by the data analysis in this study. The relationship between the relative contents of the two compounds and the C<sub>29</sub> $\alpha\alpha\alpha$ -20S/(20S+20R) of the E<sub>2</sub>s<sub>1</sub> and E<sub>2</sub>s<sub>3</sub> source rock samples is shown graphically in Fig. 7a–d. When the C<sub>29</sub> $\alpha\alpha\alpha$ -20S/(20S+20R) ratio is less than 0.4, the contents of compounds A and B in the E<sub>2</sub>s<sub>3</sub> source rock samples are relatively low, increase with the increase of the C<sub>29</sub> $\alpha\alpha\alpha$ -20S/(20S+20R) ratio is approximately 0.45, and finally decrease with the increase of the C<sub>29</sub> $\alpha\alpha$ -20S/(20S+20R) ratio is approximately 0.45, and

ratio. An identical correlation between the  $C_{29}\alpha\alpha\alpha$ -20S/(20S+20R) ratio and the contents of compounds A and B is also observed in the  $E_{2}s_{1}$  source rock samples. However, due to the lack of samples with higher thermal maturity in the  $E_{2}s_{1}$  source rock samples, when the  $C_{29}\alpha\alpha\alpha$ -20S/(20S+20R) ratio is greater than 0.5, the covariant relationship between the  $C_{29}\alpha\alpha\alpha$ -20S/(20S+20R) ratio and the contents of compounds A and B cannot be established. According to the correlation between the  $C_{29}\alpha\alpha\alpha$ -20S/(20S+20R) ratio and vitrinite reflectance ( $R_{0}$ ) established by Peters et al. (2005), when the  $C_{29}\alpha\alpha\alpha$ -20S/(20S+20R) ratios are 0.4 and 0.5, the corresponding  $R_{0}$ values are approximately 0.5% and 0.7%, respectively. The above analysis indicated that compounds A and B in source rocks began to generate in large quantities when  $R_{0}$  was 0.5%, and the concentration of these two compounds reached the maximum when  $R_{0}$  was ~0.6%. With the increase of the thermal maturity degree, the



Fig. 7. The crossplot of C<sub>29</sub>aaa-20S/(20S + 20R) versus A/C<sub>23</sub>TT and B/C<sub>23</sub>TT of E<sub>2</sub>S<sub>1</sub> (b, d) and E<sub>2</sub>S<sub>3</sub> (a, c) source rock samples in study area.

contents of these two compounds gradually decreased after  $R_0$  was greater than 0.7%.

In addition, the relationship between the contents of compounds A and B and burial depth also supports this view. The contents of compounds A and B in the  $E_{2s_3}$  source rock samples began to increase at a depth of ~3300 m, reached a maximum value at approximately 3500–3600 m, and then gradually decreased with increasing burial depth (Fig. 8a, c). The concentrations of compounds A and B in the  $E_{2s_1}$  source rocks also began to increase when the burial depth was approximately 3300 m, but the variation after the burial depth reached more than 3300 m could not be obtained due to the lack of deeper buried samples (Fig. 8b, d). According to the relationship between the burial depth and  $R_0$  of the Shahejie Formation in the Bozhong sag, the  $R_0$  values of source rocks buried at ~3300 m and ~3600 m are approximately 0.5% and 0.7%, respectively (Yu et al., 2021).

Based on the data analysis in this study, the distribution and concentration of compounds A and B are affected by the thermal maturity and biological source, while the sedimentary environment has a limited influence. The degree of thermal maturity controls the chemical bond rupture of the precursors of compounds A and B, and finally leads to the enrichment or deficiency of the two



Fig. 8. The relationship between the A/C<sub>23</sub>TT and B/C<sub>23</sub>TT ratios and burial depth of E<sub>2</sub>s<sub>1</sub>(b, d) and E<sub>2</sub>s<sub>3</sub> (a, c) source rock samples in study area.



Fig. 9. Crossplot of A/C<sub>23</sub>TT versus B/C<sub>23</sub>TT of E<sub>2</sub>s<sub>1</sub> and E<sub>2</sub>s<sub>3</sub> source rock samples in study area.

compounds in the source rocks. Generally, in the low maturity stage (0.5% <  $R_0$  < 0.7%), the concentrations of compounds A and B in source rocks are high, and their contents decrease with the increasing maturity. However, it should be noted that the biological source of organic matter also plays an important role in controlling the concentrations of these two compounds in the source rocks, which is why the concentrations of compounds A and B in the E<sub>2</sub>s<sub>1</sub> source rocks are significantly higher than those in the E<sub>2</sub>s<sub>3</sub> source rocks with the same thermal maturity degree. In this study, compound C has no obvious correlation with the sedimentary environment and thermal maturity, and its controlling factors need to be further studied.

#### 4.5. Application of unusual C<sub>24</sub> tetracyclic terpanes

Compounds A and B are closely related to the biological source and thermal maturity degree and thus are helpful for distinguishing the source rocks and crude oil samples. In this study, the high contents of compounds A and B in the E<sub>2</sub>s<sub>1</sub> source rock samples are significantly different from those in the E<sub>2</sub>s<sub>3</sub> samples, suggesting a large amount of terrestrial angiosperm input and lower thermal maturity. This finding is consistent with the basic geological information on the E<sub>2</sub>s<sub>1</sub> and E<sub>2</sub>s<sub>3</sub> source rock samples, which provides indirect evidence for the validity of these new parameters. The combination of two parameters constructed in this study A/C<sub>23</sub>TT and B/C<sub>23</sub>TT can successfully distinguish the E<sub>2</sub>s<sub>1</sub> and E<sub>2</sub>s<sub>3</sub> source rocks in the study area (Fig. 9). In addition, the  $10\beta$ (H)-des-A-oleanane/C24 des-A-hopane ratio has been effectively used for oilsource rock correlation analyses in previous studies. For example, the  $10\beta(H)$ -des-A-oleanane/C<sub>24</sub> des-A-hopane ratio combined with the oleanane/C<sub>30</sub> hopane ratio can adequately distinguish crude oil from the Zhu 1 Depression and Zhu 3 Depression (Zhu et al., 2015). Bao et al. (2018) successfully distinguished lacustrine oils and coaly oils based on a crossplot of the relative content of compounds A and C versus the bicadinane  $T/C_{30}$  hopane ratio. All these examples indicate that the parameters constructed by the three unusual compounds are expected to be effective parameters in oil source correlations.

#### 5. Conclusions

Three unusual tetracyclic terpanes, namely,  $10\beta(H)$ -des-A-

oleanane, 10β(H)-des-A-lupane and C<sub>24</sub> des-A-hopane, are identified in the E<sub>2</sub>s<sub>1</sub> and E<sub>2</sub>s<sub>3</sub> source rock samples in the Bozhong subbasin based on diagnostic ion fragments and retention time.  $10\beta(H)$ -des-A-oleanane and  $10\beta(H)$ -des-A-lupane likely originated from terrestrial angiosperms, and C<sub>24</sub> des-A-hopane likely originated from prokaryotic organisms. Abundant inputs of terrestrial angiosperms provided the material basis for the formation of compounds A and B. Thermal maturity plays an important role in controlling the formation and enrichment of compounds A and B in geological sediments. The data analysis in this study indicates that compounds A and B are largely generated and enriched in the source rocks in the low maturity stage ( $0.5\% < R_0 < 0.7\%$ ). Biological sources and thermal maturity are the main factors controlling the relative contents of compounds A and B in the studied samples. In addition, the A/ $C_{23}$ TT and B/ $C_{23}$ TT ratios are expected to be effective parameters in oil source correlations in the Bozhong subbasin.

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